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iso-PsE, a new pseudopterosin

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ABSTRACT

We describe the discovery of a new member of the pseudopterosin class of marine natural products. Its structure is isomeric with that of pseudopterosin E and has therefore been given the name iso-PsE. - 2008 Elsevier Ltd. All rights reserved.

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The pseudopterosins constitute a well-known class of more than twenty marine natural products. These interesting substances have been the focus of many investigations, some dealing with their isolation and characterization,^{[1](#page-1-0)} others with synthesis,^{[2](#page-1-0)} and others with their biological activity.^{[3](#page-1-0)} As illustrated in [Figure 1,](#page-1-0) pseudopterosins A–D (1–4) differ simply by the degree of acylation of the D -xylose sugar unit, while pseudopterosins E (5) and K (6) differ in the location of the L-fucose sugar residue and in the absolute configuration of the stereogenic centers within the tricyclic diterpene core.

In order to conduct a detailed structure–activity relationship study, we needed to develop a simple protocol whereby we could obtain substantial quantities of the tricyclic aglycon core. We began our search by using a well-established sequence wherein the phenolic hydroxyl group of raw material isolated from the feathers of Pseudopterogorgia elisabethae was converted to methyl ether

(MeI, K_2CO_3 , acetone, 60 °C, 24 h, 97%) and the sugar residue was removed (1 N HCl, MeOH, 50 °C, 4 h, 84%).^{1b,4} The specific rotation of an HPLC-purified sample of the resulting material was in accord with that reported for the aglycon derived from pseudopterosin A methyl ether (10; PsA–OMe).^{1b} We were surprised by this observation since the spectral data for the pseudopterosin that were used to begin the sequence matched those of pseudopterosin $K(6;$ PSK).^{1b} The problem, and therefore the source of our surprise, was that the aglycons of PsA and PsK (7 and 8, respectively) are enantiomers and ought to display specific rotations of equal magnitude but opposite sign. Therefore it appeared that we were dealing with a previously uncharacterized member of the pseudopterosin class of natural products, one possessing a PsA (1) core but with a sugar unit that is found in PsK (6) .

Given the fact that specific rotations do not always provide a reliable assessment of structure, 5 we felt that it was important to adopt a conservative and unambiguous approach to establish the structure of the tricyclic core. The idea was to determine whether the aglycon in question would be converted to PsA O-methyl ether (10), or to the diastereomeric structure 11 upon glycosylation with enantiomerically pure trichloroacetimidate 9.^{[6](#page-1-0)} Isolation of 10 would confirm our suspicion that the tricyclic core was of the PsA, rather than the PsK variety.

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Figure 1. Structure of selected pseudopterosins.

In practice, 2,3,4-tri-O-acetyl-β-D-xylopyranosyl trichloroacetimidate (9) was coupled to the aglycon (BF₃ etherate, 4 Å molecular sieves, CH₂Cl₂, –78 °C; 91%), and the acetate units were removed $(K_2CO_3$, MeOH, room temperature; 95%) to afford an adduct that crystallized slowly upon standing; a single crystal X-ray analysis was performed (see Scheme 1).

Knowing that the configuration at each of the stereogenic centers in the sugar portion of the structure matches those found in D-xylose, the result of the single crystal X-ray analysis revealed the product to be PsA–OMe (10) rather than the diastereomer 11 (note Scheme 1). This discovery allows us to confidently formulate **12** as the structure for the new pseudopterosin.⁷ Given its isomeric relationship to PsE (6) , we suggest the name iso-PsE $(12)^8$

Since iso-PsE is a newly characterized substance, we carried out a preliminary screening and comparison of its pharmalogical properties with other pseudopterosins. Like its relatives, iso-PsE (12)

X-ray derived structure corresponding to PsA-OMe (**10**)

Scheme 1. Transformations used to establish the structure of iso-PsE (12). For ease of visualization against the white background, the hydrogens in the X-ray derived structure for PsA–OMe (10) are shown in blue. Reagents and conditions: (i) MeI, K₂CO₃, acetone, 60 °C, 24 h (97%); (ii) 1 N HCl, MeOH, 50 °C, 4 h (84%); (iii) 9, BF₃ etherate, 4 $\,$ A mol sieve, CH₂Cl₂, -78 °C (91%); (iv) K₂CO₃, MeOH, rt (95%).

inhibits phorbol myristate acetate (PMA) induced inflammation of the inner pinnae of a mouse ear with an ED_{50} of 27 ug/ear, compared with an ED₅₀ of 8 ug/ear for PsA (1) and 22 ug/ear for PsA– OMe (10). Thus, its activity is nearly the same as 10. iso-PsE (12) also inhibits neutrophil infiltration at the application site. This is a key feature of the unique anti-inflammatory activities of the pseudopterosins in that they lack significant activity on eicosanoid biosynthesis but appear to act, in part, by blocking the release of pro-inflammatory mediators from neutrophils and macrophages.^{3a} Recent reports described a role of neutrophil infiltration in woundhealing, while recent clinical reports, although debated, have emphasized an inflammatory role of neutrophil infiltration and degranulation during injury.⁹

In addition, at the cellular level iso-PsE (12), as well as PsA (1) and PsA–OMe (10), significantly decreases basal levels of phagocytosis in cultured tetrahymena cells.¹⁰ In cultured human embryonic kidney cells (HEK-293) expressing several G-protein coupled receptors, isotope binding studies indicate that iso-PsE (12) displays a significant degree of selectivity toward adenosine A_{2A} and A_3 receptors when tested at concentrations above 5 μ M. At these concentrations iso-PsE (12), although markedly less potent than adenosine analogs, 11 followed binding kinetics similar to the adenosine standards. Studies are underway to examine effects of iso-PsE (12) in functional assays in cells expressing adenosine receptors.

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- 7. As this manuscript was being prepared, a sample of iso-PsE (12) was crystallized. The resulting single crystal X-ray analysis left no doubt that the structural assignment given herein is correct.
- 8. The spectral data for iso-PsE (12) are: ¹H NMR (400 MHz, CDCl₃) δ 1.02 (3H, d, $J = 6.0$, 1.05-1.09 (1H, m), 1.11 (3H, d, $J = 7.3$), 1.24 (3H, d, $J = 6.7$), 1.36-1.43 (1H, m), 1.50–1.57 (1H, m), 1.59–1.62 (1H, m), 1.63–1.69 (1H, m), 1.65 (3H, s), 1.74 (3H, s), 1.94–1.99 (1H, m), 1.96 (3H, s), 2.05–2.13 (2H, m), 3.39 (1H, m), 3.58 (1H, d, J = 9.2), 3.91 (1H, s), 4.02 (1H, br s, OH), 4.15 (2H, apparent s), 4.34 $(1H, q, J = 6.5)$, 4.65 $(1H, br s, OH)$, 5.06 $(1H, s)$, 5.09 $(1H, d, J = 9.2)$, 5.66 $(1H, br s)$

s, OH), 8.70 (1H, br s, OH); ¹³C NMR (100 MHz, CDCl₃) δ 11.0 (C-20), 16.4 (C-6′), 17.8 (C-16), 21.2 (C-18), 24.3 (C-19), 25.9 (C-17), 27.3 (C-7), 28.3 (C-5), 30.2 (C-
3), 30.5 (C-6), 35.7 (C-1), 39.6 (C-2), 43.0 (C-4), 67.8 (C-5'), 69.1 (C-2'), 70.7 (C-
3'), 72.4 (C-4'), 103.7 (C-1'), 121.3 (C-11), 128. electrospray in CH₃CN) m/z found 469.2540, calcd 469.2560 for C₂₆H₃₈O₆Na.

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G-protein coupled receptor (GPCR) pathway and modulation of these receptors have shown positive effects in the acceleration of wound-healing.